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Chemistry: Form WS6.4.1A Name _____ PHASES OF MATTER Date _____ Period _____ Using the Combined Gas Law If a balloon is pulled over the neck of a flask, and the setup is placed on a hot plate, the balloon blows up as it heats up. This happens even though no additional air can get into the balloon.

Using the Combined Gas Law - Evan's Regents Chemistry Corner

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Uik ,f (Continued l') e y (fpe4) >j '// t / 3 Cf Form WS6, 3. 1A nLrf: Phase Changes PHASES OF MATTER 4. Page 2 Why do substances change phase as the temperature changes? Refer to the distance between particles and the force of f Le_ attraction between them in your answer.

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Chemistry: Form WS6.5.1A Name _____ PHASES OF MATTER Date _____ Period _____ The Ideal Gas How do gases behave? In order to study gases, chemists have devised a model. The model is called an ideal gas (a gas which explains the behavior of all gases).

The Ideal Gas - Evan's Regents Chemistry Corner

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f. Form N5 . 5A CHEMICAL FORMULAS AND EQUATIONS Conservation of Mass Answer the questions below by circling the number of the correct response 1. If 46 g of X combines with 16 g of Yto form Z, how much Zis formed? (1)30g (2)2.9g (3)724g (4)62g 2. The formula mass of suite acid (H2S04) is (1) 194amu, (2)98 amu, (3)50 amu, (4)192 amu 3.

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Phases of matter change by gaining or losing energy.. Heat decides the state of matter. Adding heat or taking off heat brings in change of state/ phase. Soild -----> Liquid This is called melting process (solid needs to be heated, example melting of butter, solid on taking up heat changes to liquid)

Phases of Matter - Chemistry | Socratic

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