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11. In your own words, what is the difference between a sublevel and an orbital? Orbitals are contained within sublevels. A sublevel is a grouping of orbitals. 12. How many electrons can fit in each of the following energy levels: 1st energy level = 2 because only 2 can fit in an s sublevel and

the first energy level only has an s sublevel.

### **Information: Energy Levels and Sublevels**

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