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Name SECTION 2 Date CHAPTER 11 REVIEW Gases Class SHORT ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the

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The number 6.022 × 10 23, called Avogadro's number after the 19th-century chemist Amedeo Avogadro, is the number we use in chemistry to represent macroscopic amounts of atoms and molecules. Thus, if we have 6.022 × 10 23 Oxygen atoms, we say we have 1 mol of Oxygen atoms.

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Modern Chemistry 6 The Periodic Law CHAPTER 5 REVIEW The Periodic Law SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. ____ When an electron is added to a neutral atom, energy is (a) always absorbed.

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