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the answer. Add a small amount of solute of known mass, if it dissolves it is unsaturated. If solid settles to the bottom filter the solution and mass the remaining solid, if the mass is the same...

## **2 POGIL Saturated and Unsaturated**

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Chemistry Extension Questions Model 2  
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100	90	80	70	60	50
40	30	20	10	0	0
10	20	30	Temperature		
(°C)		Solubility (g solute/100 g water)		40	50
60	Substance A		Substance B		

12.

According to the graph in Model 2, what is the solubility of Substance A at 30 oC?

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13.

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b. Is your answer in part a consistent with your answer to Question 7: The enthalpy of fusion given in 'Model 2 for each substance is a measure of the energy (in kJ) needed to melt a mole of that substance. a. Which type of solid has the 10 highest enthalpies of fusion.' b.

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Compound Solubility Compound

Solubility  $\text{PbCl}_2$  insoluble

$\text{Pb}(\text{CH}_3\text{CO}_2)_2$  soluble KOH soluble

$\text{Mg}(\text{OH})_2$  insoluble  $\text{MgSO}_4$  soluble



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BaSO<sub>4</sub>insoluble CaCO<sub>3</sub>insoluble  
(NH<sub>4</sub>)<sub>2</sub>CO<sub>3</sub>soluble Hg<sub>2</sub>Cl<sub>2</sub>insoluble  
CaCl<sub>2</sub>soluble K<sub>2</sub>S soluble CuS insoluble  
10. When two electrolyte solutions are  
mixed, new pairings of cations and  
anions may or may not result in a  
metathetical reaction.

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## 5 - Answers Limiting ...

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14. The solubility product constant,  $K_{sp}$ , for silver carbonate,  $Ag_2CO_3$ , is  $8.10 \times 10^{-12}$ . Compare the solubility of silver carbonate in water to its solubility in a 0.100 M solution of silver nitrate.

Support your comparison with calculations similar to those in Model 1.

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POGIL: Acids,Bases,pH pOH POGIL:  $K_a$   
and  $K_b$  POGIL: Hydrolysis POGIL: Buffers  
POGIL: Titration curves POGIL: Acid Base  
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Questions 1. Using SI units of kilograms,  
meters, and seconds with these  
fundamental equations, determine the  
combination of units that define the  
following: i. The newton (N), the  
fundamental unit of force  $F = ma =$   
 $(\text{kg})(\text{m@s}^{-2}) = \text{kg@m@s}^{-2} = \text{N}$  ii.

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answers may include: erosion - puddles,  
hollowed out areas, areas that dip or are

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deposition ... As students share their  
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